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# Antioxidant activity of sparkling wines produced by *Champenoise* and *Charmat* methods

C.A. Stefenon<sup>a,b,\*</sup>, M. Colombo<sup>a</sup>, C. de M. Bonesi<sup>a</sup>, V. Marzarotto<sup>b</sup>, R. Vanderlinde<sup>a,c</sup>, M. Salvador<sup>a</sup>, J.A.P. Henriques<sup>a,d</sup>

<sup>a</sup> Instituto de Biotecnologia, Universidade de Caxias do Sul, Caxias do Sul, RS, Brazil

<sup>b</sup> Laboratório Randon Ltda, Rua Bento Gonçalves, 3365/114, 95020-412 Caxias do Sul, RS, Brazil

<sup>c</sup> Laboratório de Referência Enológica, Caxias do Sul, RS, Brazil

<sup>d</sup> Faculdade de Farmácia, Universidade Luterana do Brasil, Canoas, RS, Brazil

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# ABSTRACT

The objective of this study was to evaluate the antioxidant activity of 19 Brazilian sparkling wines produced by *Champenoise* and *Charmat* methods. All sparkling wines tested showed significant antioxidant activity, both *in vivo* and *in vitro* assays. In general, the *Charmat brut* possessed more antioxidant activity than *Charmat demi-sec* and *Champenoise* samples. In most of the sparkling wines studied, the majority compound found was gallic acid, although *trans*-resveratrol, (+)-catechin, (–)-epicathechin and procyanidins B<sub>1</sub>, B<sub>2</sub>, B<sub>3</sub> and B<sub>4</sub>, were also identified. Significant differences were observed in the concentrations of these compounds, when considering the *assemblage* used and the production methods.

The wine industry around the world uses similar oenological technologies and the wines are divided into categories, for example, in relation to sugar concentration or elaboration methods. The findings of this study would help the wineries to determine the sugar contents and time to mature (*sur lie*) appropriate for sensorial characteristics desired by the winemakers and consumers. Furthermore, the data can offer an improvement in the biological properties of the sparkling wines.

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# 1. Introduction

The process of making sparkling wines begins by obtaining the base wine from white grapes (*blanc de blancs*) or from white and red grapes (*blanc de noirs*). White base wines are obtained when fermentation takes place without contact between the must and the grape skins. The wines obtained will be red or *rosés*, depending on the time and intensity of this maceration (Hidalgo et al., 2004). The base wine is then submitted to a second fermentation, in order to produce carbon dioxide naturally by *Charmat* (in large containers) or *Champenoise* methods (in the bottle). Sparkling wines may be varietals (a single grape) or *assemblage/coupage* (two or more varieties and vintages) (Ribéreau-Gayón, Glories, Maujean, & Dubourdieu, 2003). The second fermentation can be followed by ageing of the wine with yeasts and the sparkling wines may be classified according to sugar content.

Sparkling wines are rich in phenolic compounds (Chamkha, Cathala, Cheynier, & Douillard, 2003; Ibern-Gómes et al., 2000; Pozo-Bayón, Hernández, Martín Álvares, & Polo, 2003), with known

antioxidant activity (Cartron et al., 2003; Roig, Cascón, Arola, Bladé, & Salvadó, 2002; Satué-Garcia, Andrés-Lacueva, Lamuela-Raventós, & Frankel, 1999; Yilmaz & Toledo, 2004). The content of these compounds, however, depends on several factors, including variety of grape, fruit growth and ripening conditions, quality of the base wine, yeast used and *sur lie* (time needed to mature) (Cortell, Halbleib, Gallagher, Righetti, & Kennedy, 2005; Delgado, Martín, del Álamo, & González, 2004; Mazauric & Salmon, 2005, 2006).

Several studies have already been performed in order to evaluate the antioxidant activity of red and white wines (Cartron et al., 2003; De Beer, Joubert, Gelderblom, & Manley, 2003; Jamroz & Beltowski, 2001; Landrault et al., 2001); however, there are few data on the antioxidant capacity of sparkling wines (Cartron et al., 2003; Satué-Garcia et al., 1999). Furthermore, there are no reports on the influence of the different methods for making and/or sugar concentration on the biological activity of these wines.

Therefore, the purpose of this study was to determine the antioxidant capacity of Brazilian sparkling wines *in vitro* (scavenging capacity of the free radical 1,1-diphenyl-2-picrylhydrazyl (DPPH<sup>•</sup>) and *in vivo* (in eukaryotic cells of the *Saccharomyces cerevisiae* yeast). The influence of the different methods used in manufacture (*Charmat* or *Champenoise*), the concentration of sugar (*brut* or *demisec*) and the phenolic composition on the antioxidant activity of these wines was also evaluated.





<sup>\*</sup> Corresponding author. Address: Laboratório Randon Ltda, Rua Bento Gonçalves, 3365/114, 95020-412 Caxias do Sul, RS, Brazil. Tel.: +55 54 3225 1499. *E-mail address:* claudia.alberici@terra.com.br (C.A. Stefenon).

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# 2.1. Samples

Nineteen sparkling wines were studied: 12 Charmat (seven brut and five demi-sec) and seven Champenoise, made by seven different wineries in the "Serra Gaúcha", the mountains situated in the south of Brazil. In each of these groups, the performance in relation to its respective base wine was evaluated too. The main characteristics of the sparkling wines used are shown in Table 1. These samples were made from 12 varieties: Pinot Noir (PN), Chardonnay (CH), Italian Riesling (IR), Semillon (SE), White Muscat (WMu), Merlot (ME), Cabernet Sauvignon (CS), Prosecco (PR), White Malvasia (WMa), Candia Malvasia (CMa), Canelli Muscat (CMu) and Alexandria Muscat (AM). The first fermentation (base wines) was performed at 15 °C for an average of 16 days. In the second fermentation, the mean temperature was  $12 \pm 2$  °C with the foam formation time varying between 30 and 90 days. The ageing period varied from zero to 540 days (Table 1). Except for sparkling wines 9, 10 and 11, obtained from fermentations with S. cerevisiae, in all the others S. bayanus was used in both fermentations. In order to perform the assays, the sparkling wines were previously degassed, using a vacuum pump with a valve for air removal, coupled to a workbench agitator.

# 2.2. Chemical reagents

DPPH<sup>-</sup>, *trans*-resveratrol, (+)-catechin, (–)-epicatechin, gallic acid and procyanidin  $B_3$  were acquired from Sigma–Aldrich, St. Louis, MO. The procyanidins  $B_1$ ,  $B_2$  and  $B_4$  were kindly provided by Dr. Regina Vanderlinde (Instituto Brasileiro do Vinho, Bento Gongalves, Brazil). The anthocyanins cyanidin-3-glycoside, del-phinidin-3-glycoside, peonidin-3-glycoside and malvidin-3-glyco-

#### Table 1

Main characteristics of the sparkling wines (SW) studied.

side were acquired from Extrasynthese, Genay, France. The other reagents were acquired from E. Merck, Damstadt, Germany.

#### 2.3. Oenological analysis of sparkling wines

The alcohol content, total acidity, pressure, volatile acidity, pH, free and total  $SO_2$ , dry extract and reduced dry extract, concentration of sugar and ascorbic acid were determined using the methods described by Zoecklein, Fugelsang, Gump, and Nury (2000). All analyses were performed in duplicate.

# 2.4. Determination of polyphenols by UV spectrophotometry

Total polyphenols and total hydroxycinnamates were quantified by measuring the absorbances at 280 and 320 nm (UV-1700 spectrophotometer, Shimadzu, Kyoto, Japan), respectively. The results of total polyphenols were expressed as mg/l of catechin and those of hydroxycinnamates as mg/l of caffeic acid. The total flavonoids (*TF*) were calculated using the following formula, as described by Iland, Ewart, Sitters, Markides, and Bruer (2000):

$$TF = [(A_{280} - 4) - 0.66] \times (A_{320} - 1.4).$$

The results were expressed in mg/l of catechin. All the analyses were performed in duplicate.

#### 2.5. Determination of polyphenols by HPLC

A 5 ml aliquot of each sample was filtered through a cellulose membrane with a 0.20  $\mu$ m diameter just before the analysis of the major phenolic compounds by high performance liquid chromatography using a Hewlett–Packard (Palo Alto, CA) 1100 series

SW	Assemblage	Wineries	<i>Sur lie</i> 1 (days)	Ascorbic acid ± SD <sup>2</sup> (mg/l)	Sugar ± SD (g/l)	TP <sup>3</sup> (mg/l of C <sup>6</sup> ) ± SD	TF <sup>4</sup> (mg/l of C) ± SD	THC <sup>5</sup> (mg/l of $CA^7$ ) <sup>*</sup> ± SD
Charm	at brut		(	(8)-)	(81-)	- ,	-)	
1	40% PN, 60% CH	A*	30	33.3 ± 0.03 <sup>a**</sup>	$12.5 \pm 0.23^{a}$	$354 \pm 4.19^{a}$	$156 \pm 2.94^{a}$	$39.8 \pm 0.22^{a}$
2	60% PN, 40% CH	В	126	$21.9 \pm 0.01^{b}$	$5.48 \pm 0.26^{b}$	$185 \pm 7.13^{b}$	$52.3 \pm 4.19^{b}$	$26.6 \pm 0.61^{b}$
3	50% IR, 30% SE, 20% WMu	C	No ageing	$28.9 \pm 0.03^{\circ}$	$11.4 \pm 0.08^{\circ}$	$163 \pm 7.13$ $164 \pm 4.19^{\circ}$	6.16 ± 1.68 <sup>c</sup>	$31.5 \pm 0.50^{\circ}$
4	50% IR, 33% PN, 17% CH	В	60	$11.4 \pm 0.04^{d}$	$9.33 \pm 0.20^{d}$	$193 \pm 1.68^{b}$	$23.8 \pm 1.67^{d}$	$33.9 \pm 0.55^{d}$
5	30% IR, 70% CH	D	60	$21.8 \pm 0.03^{b}$	$8.94 \pm 0.17^{d}$	$520 \pm 0.00^{d}$	$302 \pm 1.67^{e}$	$43.7 \pm 0.33^{\circ}$
6	100% CH	E	30	$22.6 \pm 0.06^{b}$	$10.5 \pm 0.22^{e}$	$420 \pm 3.35^{\circ}$	$220 \pm 12.15^{\text{f}}$	$40.1 \pm 3.05^{a}$
7	40% IR, 10% ME, 50% CS	В	193	$32.6 \pm 0.04^{a}$	$10.6 \pm 0.29^{e}$	$1350 \pm 3.35^{\rm f}$	$780 \pm 6.70^{g}$	$113 \pm 0.66^{f}$
Chami	penoise							
8	40% PN, 60% CH	А	180	$11.9 \pm 0.03^{d}$	$5.84 \pm 0.14^{b}$	$362 \pm 0.84^{a}$	$192 \pm 1.67^{h}$	$34.1 \pm 0.22^{d}$
9	100% PR	F	150	$20.5 \pm 0.02^{e}$	$7.28 \pm 0.15^{f}$	$483 \pm 5.86^{g}$	$234 \pm 5.87^{f}$	$50.0 \pm 0.05^{g}$
10	100% CH	F	540	$16.7 \pm 0.03^{f}$	$8.04 \pm 0.19^{d}$	$620 \pm 4.19^{h}$	$365 \pm 4.19^{i}$	$51.3 \pm 0.00^{h}$
11	20% PN, 80% CH	F	270	$21.2 \pm 0.01^{b}$	$7.24 \pm 0.11^{f}$	493 ± 2.93 <sup>g</sup>	$242 \pm 5.03^{f}$	$50.4 \pm 0.39^{g}$
12	10% CH, 60% ME, 30% PN	F	365	$79.4 \pm 0.02^{g}$	$9.91 \pm 0.07^{d}$	$2790 \pm 41.90^{i}$	1870 ± 22.63 <sup>j</sup>	185 ± 3.89 <sup>i</sup>
13	50% PN, 50% CH	F	150	$43.6 \pm 0.04^{h}$	$6.59 \pm 0.20^{g}$	616 ± 37.71 <sup>h</sup>	301 ± 21.37 <sup>e</sup>	63.3 ± 3.33 <sup>j</sup>
14	20% PN, 80% CH	G	365	46.5 ± 0.03 <sup>i</sup>	$5.89 \pm 0.19^{b}$	$375 \pm 0.84^{j}$	$216 \pm 0.42^{f}$	31.8 ± 0.11 <sup>c</sup>
Charm	at demi-sec							
15	50% IR, 30% SE, 20% WMu	С	No ageing	29.0 ± 0.03 <sup>c</sup>	$37.5 \pm 0.39^{h}$	$185 \pm 0.84^{b}$	$15.4 \pm 1.68^{k}$	$34.2 \pm 0.16^{d}$
16	30% IR, 70% CH	D	60	$19.1 \pm 0.07^{j}$	$51.1 \pm 0.14^{i}$	$509 \pm 3.35^{k}$	$306 \pm 9.22^{e}$	$42.8 \pm 0.44^{e}$
17	100% CH	Е	30	$19.5 \pm 0.04^{j}$	$54.4 \pm 0.09^{j}$	523 ± 3.35 <sup>d</sup>	297 ± 1.68 <sup>e</sup>	$45.4 \pm 0.33^{k}$
18	56% WMa, 25% CMa, 10% CMu, 9% AM	В	8	$61.3 \pm 0.03^{k}$	$36.2 \pm 0.13^{k}$	$340 \pm 0.84^{1}$	$146 \pm 0.42^{1}$	$38.9 \pm 0.05^{a}$
19	74% IR, 14% PN, 12% CH	В	30	15.7 ± 0.03 <sup>1</sup>	$36.4 \pm 0.04^{k}$	$217 \pm 0.42^{m}$	$32.6 \pm 0.42^{m}$	$36.9 \pm 0.05^{1}$

<sup>1</sup> Time needed to mature.

<sup>2</sup> Standard deviation.

<sup>3</sup> Total polyphenols (TP).

<sup>4</sup> Total flavonoids (*TF*).

<sup>5</sup> Total hydroxycinnamates (THC).

<sup>6</sup> Catechin (C).

<sup>7</sup> Caffeic acid (CA).

<sup>\*</sup> Distinct letters corresponding to different wineries in the Serra Gaúcha/Rio Grande do Sul/Brazil.

<sup>\*\*</sup> Data followed by different letters for each column differ significantly by Kruskal–Wallis *H* test ( $p \leq 0.05$ ).

LC liquid gradient, with a Diode Array Detector (DAD). A Zorbax 300 SB C18 ( $12 \text{ mm} \times 4.6 \text{ mm} \times 5 \mu \text{m}$ ) pre-column and a C18-ODS ( $150 \text{ mm} \times 4 \text{ mm} \times 5 \mu \text{m}$ ) (Agilent Technologies, Santa Clara, CA) column were used. The specific phenols quantified were *trans*-resveratrol (Jeandet et al., 1995), anthocyanidins (OIV – Resolution OENO 22/2003), procyanidins B<sub>1</sub>, B<sub>2</sub>, B<sub>3</sub> and B<sub>4</sub>, (+)-catechin, (–)-epicatechin and gallic acid (Lamuela-Raventós & Waterhouse, 1994).

# 2.6. Evaluation of antioxidant activity in vitro

The scavenging capacity of free radical DPPH<sup>·</sup> was measured by adding to the sparkling wines, pure or diluted in distilled water [0.1%; 1.0%; 10% and 50% (v/v)], a tris–HCl buffer solution (100 mM, pH 7.0) containing 250  $\mu$ M of DPPH<sup>·</sup> dissolved in ethanol. In the control tube, sterilised distilled water was used in lieu of sparkling wines. The tubes were kept in the dark for 20 min and the absorbance was measured at 517 nm (Shimadzu UV-1700 spectrophotometer) (Yamaguchi, Takamura, Matoba, & Terão, 1998). The results were expressed in values of  $IC_{50}$  (quantity of sparkling wine needed to reduce 50% of the free radical DPPH<sup>·</sup>), calculated by polynomial regression graphs (Mensor et al., 2001), using the mean of triplicates.

#### 2.7. Evaluation of antioxidant activity in vivo

The assays *in vivo* were performed using cells of *S. cerevisiae* XV 185-14c yeast (MAT  $\alpha$  ade 2-1, arg 4-17, his 1-7, lys 1-1, trp 1-1, trp 5-48, hom 3-10), kindly provided by Dr. R.C. Von Borstel (Department of Genetics, University of Alberta, Canada). Cell suspensions of  $2 \times 10^6$  cells/ml obtained from the exponential growth phase were treated with hydrogen peroxide (75 mM), in the presence and absence of sparkling wines. The tubes were incubated for 1 h at 28 °C. Then the samples were diluted in a 0.9% (w/v) sodium chloride solution, seeded into a YPD culture medium (10 g/l of yeast extract, 20 g/l of peptone, 20 g/l of dextrose and 20 g/l of agar–agar) and incubated for 72 h at 28 °C. After incubation, the colonies were counted, and 100% survival was considered the total of colonies observed on the control plate (untreated cells) (Wilmsen, Spada, & Salvador, 2005).

# 2.8. Data analysis

The data were analysed using the following tests: Kruskal–Wallis *H*, Spearman Correlation and Principal Components Analysis (PCA), using SPSS 12.0 for Windows (SPSS Inc., Chicago, IL).

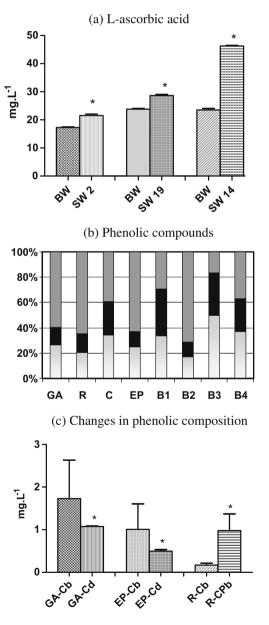
# 3. Results and discussion

# 3.1. Oenological analysis and determination of polyphenols by UV spectrophotometry

The alcohol contents of the different sparkling wines analysed varied from 11.23% to 13.05% (v/v), and total acidity from 5.08 to 8.21 g/l of tartaric acid. The mean levels of pressure, volatile acidity and pH were  $5.7 \pm 0.2$  atm,  $0.588 \pm 0.091$  g/l of acetic acid and  $3.29 \pm 0.14$ , respectively. The mean concentration of free SO<sub>2</sub> was  $20.0 \pm 8.12$  mg/l and total SO<sub>2</sub> was  $122 \pm 37.3$  mg/l. The values were below the allowed level for volatile acidity, free SO<sub>2</sub> and total SO<sub>2</sub>, indicating that the grapes were healthy and that good vinification practices were used (Boulton, Singleton, Bisson, & Kunkee, 1995). The analysis of the dry extract and reduced dry extract showed, respectively, values of  $23.5 \pm 3.57$  and  $19.1 \pm 2.95$  mg/l for the *brut* samples, and  $58.3 \pm 9.93$  and  $16.1 \pm 2.69$  mg/l for the *demi-sec* sparkling wines (data not shown). The sugar concentra-

tion varied from 5.48 to 12.5 g/l, for the *brut* samples, and from 36.2 to 54.4 g/l for the *demi-sec* sparkling wines (Table 1). These values are within the range established by Brazilian law (Brasil, 1990) for sparkling wines.

The ascorbic acid content of the sparkling wines studied varied from 11.4 to 79.4 mg/l (Table 1). Up to the present, there are few data in the literature concerning the content of this acid in sparkling wines. We know that its concentration is the result of the variety of grape, degree of maturity (Ribéreau-Gayón et al., 2003) and the amount of sunlight on the vine (Valpuesta & Botella, 2004), for example. Fig. 1a shows that sparkling wines had an increase in ascorbic acid concentration, in comparison to the base wine (samples **2**, **15** and **19**). Although the use of the ascorbic acid in wines is a well-known practice (Marks & Morris, 1993), we believe that this result is probably due to yeast metabolism (Hancock, Galpin, & Viola, 2000; Sauer, Branduardi, Valli, & Porro, 2004; Smir-



**Fig. 1.** Profiles of L-ascorbic acid (a) in base wine (BW) and respective sparkling wine (SW), polyphenols (b) by HPLC:  $\Box$ ; Charmat brut (Cb),  $\Box$ ; Charmat of demi-sec (Cd), gallic acid (GA), trans-resveratrol (R), catechin (C), epicatechin (EP), and procyanidins (B<sub>1</sub>-B<sub>2</sub>-B<sub>3</sub>-B<sub>4</sub>); and changes in phenolic composition (c) mediated by *sur lie* and sugar concentration. (a) L-Ascorbic acid, (b) phenolic compounds, and (c) changes in phenolic composition.

noff, Conklin, & Loewus, 2001), as there was no ascorbic acid addition to the sparkling wines assayed. Furthermore, in sample **14**, for which the *sur lie* period was 360 days (Table 1), the concentration of vitamin C practically doubled. New tests to evaluate the biosynthesis of ascorbic acid by oenological yeasts are currently being carried out as a result.

Higher values of total polyphenols, total flavonoids and total hydroxycinnamates were found, obviously, in red sparkling wines (samples 7 and 12). As for the whites, major variations were observed, depending partially on the method by which the sparkling wine was made (Table 2). Sample 14 (Champenoise) showed a higher concentration of total polyphenols and total flavonoids compared to the base wine of origin. On the other hand, the concentration of these compounds diminished in Charmat samples (both *brut* and *demi-sec*) compared to their respective base wines (Fig. 1b). The mean reduction of total polyphenols and total flavonoids levels observed in *Charmat* sparkling wines was  $24.58 \pm 0.72\%$  and  $57.19 \pm 4.22\%$ , respectively. The concentration of total hydroxycinnamates diminished after the second fermentation, independent of the method by which they were made (Fig. 2), probably due to the action of yeasts, which can metabolise and/or adsorb up to 20% of the content of these compounds (Ribéreau-Gayón et al., 2003; Zoecklein et al., 2000).

#### 3.2. Polyphenols analysis by HPLC

Fig. 1b shows the percentile differences on the phenolic profile of all sparkling wines. Some specific differences have been noted: sample **12** had the highest contents of all polyphenols analysed; larger amounts of *trans*-resveratrol, (+)-catechin and procyanidins  $B_1$ ,  $B_3$  and  $B_4$  were observed in sample **7**; higher values of (-)-epicatechin were obtained in sample **1** and of procyanidin  $B_2$  in sample **10**; in all sparkling wines analysed, the main phenolic component was gallic acid (data not shown).

#### Table 2

DPPH<sup>.</sup> mean values for the different sparkling wines analysed and mean survival values of the *Saccharomyces cerevisiae* yeast treated with hydrogen peroxide ( $H_2O_2$ ) 75 mM in presence and absence of different sparkling wines.

Sparkling wines	DPPH <sup>.</sup>		Survival	
	$IC_{50}^{*} \pm SD^{**}$	Rank <sup>#</sup>	±SD (%)	Rank
Charmat brut				
1	11.80 ± 1.28 <sup>*</sup>	2	93.55 ± 0.25 <sup>*</sup>	1
2	13.03 ± 1.19 <sup>b</sup>	3	93.65 ± 4.75 <sup>*</sup>	1
3	$1.83 \pm 0.49^{\circ}$	1	96.90 ± 3.10 <sup>a</sup>	1
4	$80.76 \pm 2.75^{d}$	8	$100.00 \pm 0.00^{a}$	1
5	$20.76 \pm 5.12^{e}$	4	$78.70 \pm 3.70^{b}$	3
6	$20.26 \pm 0.32^{e}$	4	96.75 ± 0.35"	1
7	$31.50 \pm 0.42^{f}$	5	$62.43 \pm 8.30^{\circ}$	4
Champenoise				
8	9.05 ± 1.04 <sup>*</sup>	2	84.22 ± 1.91 <sup>d</sup>	2
9	19.63 ± 0.63 <sup>e</sup>	4	56.80 ± 1.00 <sup>e</sup>	5
10	$30.83 \pm 2.01^{f}$	5	53.58 ± 0.95 <sup>e</sup>	5
11	11.09 ± 1.81 <sup>*</sup>	2	$34.80 \pm 0.50^{\rm f}$	6
12	$32.83 \pm 0.13^{f}$	5	$85.50 \pm 0.90^{d}$	2
13	$16.77 \pm 0.54^{b}$	3	65.41 ± 1.21 <sup>c</sup>	4
14	39.53 ± 1.53 <sup>g</sup>	6	$85.05 \pm 3.35^{d}$	2
Charmat demi-sec				
15	26.87 ± 0.58 <sup>e</sup>	4	65.45 ± 1.05 <sup>c</sup>	4
16	$32.56 \pm 0.43^{f}$	5	$64.85 \pm 2.35^{\circ}$	4
17	$25.04 \pm 0.13^{e}$	4	$79.20 \pm 3.60^{b}$	3
18	23.73 ± 0.57 <sup>e</sup>	4	$100.00 \pm 0.00^{a}$	1
19	26.10 ± 1.93 <sup>e</sup>	4	$95.30 \pm 4.70^{a}$	1
Catechin (control)	63.68 ± 1.42 <sup>h</sup>	7		
$H_2O_2$ (control)			$27.95 \pm 0.25^{g}$	7

<sup>\*</sup> *IC*<sub>50</sub> (% of amount of samples necessary to scavenge 50% of DPPH<sup>•</sup>).

\*\* Standard deviation.

<sup>#</sup> Rank in crescent order according to statistical significance (Kruskal–Wallis *H* test,  $p \le 0.05$ ) among the values of each parameter.

So far no studies have been performed on the phenolic composition of *Charmat* sparkling wines. As to the *Champenoise*, it was observed that the Brazilian sparkling wines possessed similar phenolic profiles to those reported for Spanish sparkling wines (Ibern-Gómes et al., 2000; Pozo-Bayón et al., 2003; Satué-Garcia et al., 1999), as well as in French Champagnes (Chamkha et al., 2003). To our knowledge this is the first report on the presence of procyanidins  $B_1$ ,  $B_2$  and  $B_4$  in sparkling wines.

For the red and *rosé* sparkling wines the contents of four important anthocyanins found in red grapes and wine (Zoecklein et al., 2000) were also quantified, and as expected, sample **13** (*rosé*) possessed lower concentrations among the four compounds analysed compared to samples **7** and **12** (red). The main and most plentiful anthocyanin found in red varieties, malvidin monoglycoside (Ribéreau-Gayón et al., 2003), was the main compound in the three sparkling wine samples evaluated (data not shown).

Fig. 2 shows the phenolic profile of the samples assayed. The second fermentation of base wines increased the concentration of gallic acid in the *brut* sparkling wines, probably due to the hydrolysis of procyanidins that are esterified with this acid (Jordão, 2000; Stevens et al., 2002). Conversely, in *demi-sec* sparkling wines this type of reaction seems to be a disadvantage, suggesting that sugar concentration affects the phenolic composition. The same relationship was verified on (–)-epicatechin values. A higher concentration in *brut* (*Charmat* or *Champenoise*) sparkling wines than in their respective base wines were observed (Fig. 2), and no changes were observed in the *demi-sec* sparkling wines compared to the respective base wines.

The trans-resveratrol contents (Fig. 1b) were lower in the Charmat sparkling wines (both brut and demi-sec), compared to the base wines, probably due to the finishing stages, such as clarification and or filtration (Threlfall, Morris, & Mauromoustakos, 1999; Vrhovsek, Wendelin, & Eder, 1997). With the Champenoise method, however, there was a higher content of this compound than in its base wine. Both the malolactic fermentation (Pezet & Cuenat, 1996) and the action of veasts with over expression of the  $\beta$ -glycosidase enzyme (Vrhovsek et al., 1997) may increase the concentration of *trans*-resveratrol in wines. Furthermore, sample 5 was obtained with a longer time sur lie compared to the other samples (Table 1). A positive correlation was observed between the concentration of *trans*-resveratrol and the *sur lie* time (r = 0.456; p = 0.05), suggesting an effect of contact time between the yeasts and the sparkling wine in the trans-resveratrol concentration. New assays to evaluate the  $\beta$ -glycosidase performance during the vinification are currently being carried out.

Comparing the base wines, it is observed that the (+)-catechin concentration changes in relation to the sur lie (Table 1). Samples 3, 15 and 19, elaborated with minimal sur lie times did not show differences. For a medium sur lie (sample 2), the values of this phenolic compound increased. Conversely, for greater periods of sur lie, the (+)-catechin levels were lower (sample 14). A negative correlation was found between the reduced dry extract content (which had a direct relationship with the yeast metabolism) and the concentration of (+)-catechin (r = -0.619; p = 0.01). Procyanidin B<sub>1</sub>, B<sub>2</sub>, B<sub>3</sub> and B<sub>4</sub> contents were lower after the second fermentation (data not shown). It is possible that, as seen in red wines, this reduction is due to condensation reactions, e.g., of procyanidins with proteins and polysaccharides, to the polymerisation reactions among the different procyanidins and to oxidative degradation phenomena (Lopez-Toledano, Mayen, Merida, & Medina, 2002; Ribéreau-Gayón et al., 2003). These interactions may be influenced by the different techniques adopted during the manufacturing/maturing process (Boulton et al., 1995; Flanzy, 2003; Mazauric & Salmon, 2005; Ribéreau-Gayón et al., 2003), accounting, at least in part, for the differences found.

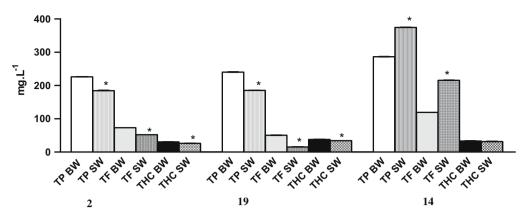


Fig. 2. Level of total polyphenols (TP), total flavonoids (TF) and total hydroxycinnamates (THC) in base wines (BW) (2, 19 and 14) and in their corresponding sparkling wine (SW) Charmat brut (2), Charmat demi-sec (19) and Champenoise (14).

#### 3.3. Evaluation of antioxidant activity in vitro

Table 3 shows that sample **3**, which was elaborated without *sur lie* (Table 1) possessed the highest degree of antioxidant activity  $(IC_{50} = 1.83 \pm 0.49\%)$ . Furthermore, sparkling wines **6** (*Charmat*) and 10 (Champenoise) were prepared exclusively with the Chardonnay variety, and the greatest capacity for scavenging of free radical DPPH<sup>•</sup> (Table 3) was observed in sample 6, which was produced with a small sur lie, and possessed a higher ascorbic acid content (Table 1) and a higher concentration of major phenolic compounds analysed by HPLC than sample 10 (Table 2).

Sparkling wines 11 and 14 were prepared using the Champenoise method by different wineries. Sample 11 possessed a greater capacity to scavenge free radical DPPH<sup>•</sup> than sample **14** (Table 3). The latter showed lower (+)-catechin, (-)-epicatechin, gallic acid and procyanidin  $B_1$ ,  $B_2$  and  $B_4$  contents (Table 2), indicating the influence of the vinification techniques (as for example the sur lie period) adopted by each winery on the phenolic composition and antioxidant activity of the sparkling wines. Interestingly, the similar assemblage (40% Pinot Noir and 60% Chardonnay) submitted to different methods of vinification in the same winery showed small differences in the antioxidant capacity. Sample 8 sur lie (Table 1; Champenoise), which presented higher antioxidant activity than sparkling wine 1 (Charmat) was only a little greater (Table 2), suggesting one more time the influence of this technique. Higher values of total flavonoids and procyanidins  $B_2$  and  $B_3$  (Table 2) were found in sparkling wine 8. New assays about the influence of the sur lie time on the antioxidant activity and phenolic profile are being currently carried out. Beyond sur lie, the sugar concentration also influenced results.

Table 3	
Variance explained by the first principal components (PC	).

РС	Eigenvalue	Explained variance (%)	Cumulative variance (%)				
Charmat brut							
1 – <i>TF</i> <sup>a</sup>	11.83	51.45	51.45				
2 – DE <sup>b</sup>	4.70	20.44	71.89				
Champenoise							
1 – Sur lie	15.94	69.29	69.29				
2 – TP <sup>c</sup>	3.92	17.03	86.32				
Charmat demi-sec							
1 – Sugar	12.41	53.95	53.95				
2 – Alcohol	4.37	19.00	72.95				

<sup>a</sup> Total flavonoids.

<sup>b</sup> Dry extract.

<sup>c</sup> Total polyphenols.

Samples of *Charmat brut* **3**, **5** and **6** showed a higher antioxidant capacity than their respective demi-sec peers (samples 15, 16 and 17) (Table 2). Higher values of (+)-catechin, procyanidin B<sub>2</sub> and gallic acid (Table 1) were found in these brut sparkling wines, compared to their *demi-sec* peers.

Apparently, antioxidant activity does not depend exclusively on the total polyphenols content. Samples 7 and 12 (red sparkling wines) with significant amounts of phenolic compounds (Tables 1 and 2) did not show the highest antioxidant activity. Studies have already demonstrated that the biological activity of resveratrol, specifically the inhibition of the tyrosinekinase protein p56, is diminished when glycosilation of the hydroxyl groups occurs (Soleas, Diamandis, & Goldberg, 1997). Therefore, it is possible that some polyphenols, when connected to carbohydrates (for instance the anthocyanidins of red wines) possess less antioxidant activity than their respective aglycones, which might account at least in part for the results observed. Among the red sparkling wines, the presence of a negative correlation between the antioxidant capacity in vitro and the concentrations of cvanidin-3-glycoside, peonidin-3-glycoside and malvidin-3-glycoside (all with a value of r = 0.985) and delphinidin-3-glycoside (r = 1), at a level of significance of p = 0.01 corroborates this hypothesis. New assays examining the sugar influence on antioxidant activity are being currently carried out.

#### 3.4. Evaluation of the antioxidant activity in vivo

In order to determine antioxidant activity in vivo, the highest non-cytotoxic concentration of sparkling wines was used, i.e., 10.0% (v/v) (data not shown). All the samples evaluated were able to protect the yeast cells against damage caused by hydrogen peroxide (Table 3).

Of the two red sparkling wines studied, sample 12, which possessed the highest polyphenol and ascorbic acid contents of all the sparkling wines evaluated (Tables 1 and 2), showed higher antioxidant activity in vivo than sparkling wine 7 (Table 2).

Differences in the antioxidant activities of sparkling wines because of the Charmat and Champenoise methods used to make them are shown in this study. Similarly to what was observed in vitro, the Charmat brut sparkling wines possessed on average a higher antioxidant capacity  $(88.85 \pm 1.79)$  than those prepared by the Champenoise method ( $60.77 \pm 3.31$ ), including the cases in which the sparkling wines were made using the same varieties/assemblages and/or by the same wineries (i.e., samples 1 and 8, and 6 and 10).

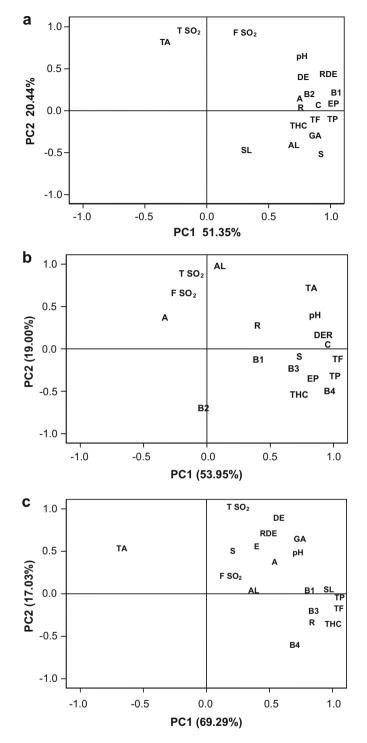
The greatest differences between the two methods of making sparkling wines are the sur lie time and the area of contact between the sparkling wine, the yeasts, vinary containers and sugar concentration. It has already been demonstrated that the yeasts are able to adsorb different compounds present in the wines, including catechin and epicatechin (Mazauric & Salmon, 2005; Ribéreau-Gayón et al., 2003). The samples of *Charmat brut* (1 and 6) prepared with a shorter sur lie time than their respective peers Champenoise (samples 8 and 10) (Table 1), possessed higher (+)-catechin and (-)-epicatechin contents (Table 2). A negative correlation was observed between the sur lie period and antioxidant activity in vivo (r = -0.519; p = 0.01), i.e., the longer the sur lie period, the smaller was the antioxidant capacity of the sparkling wines. This correlation was even greater in the group of Champenoise sparkling wines (r = -0.842; p = 0.05). The statistical analysis of the principal components (PCA), for all sparkling wines, revealed that the first two principal components explain more than 70% of the total variance (Table 3). This analysis corroborated our data, since the sur lie variable appears as one of the factors of biggest influence on the antioxidant activity in the Champenoise (Fig. 3a), when compared with Charmat sparkling wines (Fig. 3b and c). Furthermore, the Charmat demi-sec sparking wines (samples 15, 16 and 17) possessed lower antioxidant activity (Table 2) than observed for their respective *Charmat brut* peers (samples **3**, **5** and **6**), similarly to what was observed in vitro. Negative correlations between the in vivo antioxidant capacity and the sugar contents (r = -0.476; p = 0.01) and dry extract (r = -0.346; p = 0.05) were found in these two groups of sparkling wines. This is the first time that differences are shown in the antioxidant potential of sparkling wines as a function of its sugar concentration. Orange juices with added sugar showed less antioxidant activity than juices without added sugar (Franke et al., 2004), corroborating the results found in our work. The PCA analysis showed that, for the group of *Charmat demi-sec* sparkling wines, the sugar concentration was one of the main variables that had influenced the antioxidant response of these wines (Fig. 3b).

The phenolic compounds have an acknowledged antioxidant capacity, and can scavenger free radicals, chelate metals, and diminish lipid peroxidation (Cartron et al., 2003; Jamroz & Beltowski, 2001; Roig et al., 2002; Yilmaz & Toledo, 2004). Polyphenol formation by the vine is influenced by natural factors, such as variety of grape, genetic susceptibility to diseases, climate and soil, besides viticultural management (rootstock, vigour, exposure to sunlight as a function of the conduction system, fertilisation, etc.; Cortell et al., 2005; Delgado et al., 2004; Ribéreau-Gayón et al., 2003). Oenology also plays an important role in determining the phenolic profile of the final product, using techniques such as industrial maturing, press yield, maceration time and temperature, yeast used, fermentation period, clarification, stabilisation, filtration, maturing, ageing, etc. (Mazauric & Salmon, 2005; Ribéreau-Gayón et al., 2003).

In this study, the phenolic compounds showed an important role in the antioxidant activity. Interestingly, the PCA analysis showed a stronger association between the phenolic compounds and the antioxidant activity for the *Champenoise* than *Charmat* sparkling wines (Fig. 3). Therefore, the role of phenolic compounds may be greater in the antioxidant activity of the sparkling wines made by the *Champenoise* method.

Summarising, the results presented in this study show that: (a) the *Charmat* and *Champenoise* sparkling wines, both *brut* and *demisec*, possess significant antioxidant activity, which is associated with the presence of phenolic compounds; (b) *Charmat brut* sparkling wines possess higher antioxidant activity than *demi-sec Charmat* and *Champenoise*; and (c) there are major differences in the concentrations of phenolic compounds in sparkling wines as a function of the type of grape/*assemblage* used and the method used for manufacture.

New hypotheses about sparkling wine antioxidant activity, in relation to phenolic composition, were discovered with our tests



**Fig. 3.** Scores plots (PC1 vs. PC2) of the phenolic compounds and main characteristics of sparkling wines (a) *Champenoise*, (b) *Charmat demi-sec* and (c) *Charmat brut* (pH, *trans*-resveratrol (R), t-ascorbic acid (A), sugar concentration (S), catechin (C), procyanidins ( $B_1$ - $B_2$ - $B_3$ - $B_4$ ), gallic acid (GA), epicatechin (EP), total polyphenols (TP), total flavonoids (*TF*), total hydroxycinnamates (THC), dry extract (DE), reduced dry extract (RDE), total acidity (TA), *sur lie* (SL), alcohol (AL) and free and total SO<sub>2</sub> (FSO<sub>2</sub>-TSO<sub>2</sub>)).

and they will be evaluated individually. However, data shown in this work suggest that it is possible to obtain a specific phenolic profile in sparkling wines by the oenological practices adopted, thus enabling the production of sparkling wines with greater antioxidant activity and thus with a higher added value. Furthermore, the moderate/guided consumption of sparkling wines may be a positive choice in seeking a healthy life.

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#### References

- Boulton, R. B., Singleton, V. L., Bisson, L. F., & Kunkee, R. E. (1995). Principles and practices of winemaking. Nova York: International Thomson Publishing. p. 603.
- Brasil (1990). Ministério da Agricultura. Decreto n. 99.066. Regulamenta a Lei n. 7678, de 08 de novembro de 1988, que dispõe sobre a produção, circulação e comercialização do vinho e derivados do vinho e da uva. Diário Oficial da União. Brasília, 08 de março de 1990.
- Cartron, E., Fouret, G., Carbonneau, M.-A., Lauret, C., Michel, F., Monnier, L., et al. (2003). Red-wine beneficial long-term effect on lipids but not on antioxidant characteristics in plasma in a study comparing three types of wine – Description of two O-methylated derivatives of gallic acid in humans. Free Radical Research, 37, 1021–1035.
- Chamkha, M., Cathala, B., Cheynier, V., & Douillard, R. (2003). Phenolic composition of *champagnes* from chardonnay and pinot noir vintages. *Journal of Agricultural* and Food Chemistry, 51, 3179–3184.
- Cortell, J. M., Halbleib, M., Gallagher, A. V., Righetti, T. L., & Kennedy, J. A. (2005). Influence of vine vigor on grape (*Vitis vinifera* L. Cv. pinot noir) and wine proanthocyanidins. *Journal of Agricultural and Food Chemistry*, 53, 5798–5808.
- De Beer, D., Joubert, E., Gelderblom, W. C. A., & Manley, M. (2003). Antioxidant activity of South African red and white cultivar wines: Free radical scavenging. *Journal of Agricultural and Food Chemistry*, 51, 902–909.
- Delgado, R., Martín, P., del Álamo, M., & González, M.-R. (2004). Changes in the phenolic composition of grape berries during ripening in relation to vineyard nitrogen and potassium fertilisation rates. *Journal of Agricultural and Food Chemistry*, 84, 623–630.
- Flanzy, C. (2003). Enología: Fundamentos científicos y tecnológicos (2nd ed.). Madrid: AMV, Ediciones Mundi-Prensa. p. 797.
- Franke, S. I. R., Ckless, K., Silveira, J. D., Rubensam, G., Brendel, M., Erdtmann, B., et al. (2004). Study of antioxidant and mutagenic activity of different orange juices. *Food Chemistry*, 88, 45–55.
- Hancock, R. D., Galpin, J. R., & Viola, R. (2000). Biosynthesis of 1-ascorbic acid (vitamin C) by Saccharomyces cerevisae. FEMS Microbiology Letter, 186, 245–250.
- Hidalgo, P., Pueyo, E., Pozo-Bayón, M. A., Martínez Rodríguez, A. J., Martín-Álvarez, P., & Polo, M. C. (2004). Sensory and analytical study of rosé sparkling wines manufactured by second fermentation in the bottle. *Journal of Agricultural and Food Chemistry*, 52, 6640–6645.
- Ibern-Gómes, M., Andrés-Lacueva, C., Lamuela-Raventós, R. M., Buxaderas, S., Singleton, V. L., & de la Torre-Boronat, M. C. (2000). Browning of *Cava* (sparkling wine) during aging in contact with less due to the phenolic composition. *American Journal Enology and Viticulture*, 51, 30–36.
- Jamroz, A., & Beltowski, J. (2001). Antioxidant capacity of selected wines. Medical Science Monitor, 7, 1198–1202.
- Jeandet, P., Bessis, R., Maume, B. F., Meunier, P., Peyron, D., & Trollat, P. (1995). Effect of enological practices on the resveratrol isomer content of wine. *Journal of Agricultural and Food Chemistry*, 43, 316–319.
- Jordão, A. M. (2000). Estrutura e composição das proantocianidinas da uva. Evolução ao longo da maturação. Revista do Instituto Superior Politécnico de Viseu. 19, 14.
- Lamuela-Raventós, R. M., & Waterhouse, A. (1994). A direct HPLC separation if wine phenolics. American Journal of Enology and Viticulture, 45, 1–5.
- Lopez-Toledano, A., Mayen, M., Merida, J., & Medina, M. (2002). Yeast-induced inhibition of (+)catechin and (-)epicatechin degradation in model solutions. *Journal of Agricultural and Food Chemistry*, 50, 1631–1635.

- Iland, P., Ewart, A., Sitters, J., Markides, A., & Bruer, N. (2000). Techniques for chemicals analysis and quality monitoring during winemaking. Campbeltown: Patrick Iland Wine Promotions. p. 111.
- Landrault, N., Poucheret, P., Ravel, P., Gasc, F., Cros, G., & Teissedre, P.-L. (2001). Antioxidant capacities and phenolics levels of French wines from different varieties and vintages. *Journal of Agricultural and Food Chemistry*, 49, 3341–3348.
- Marks, A. C., & Morris, J. R. (1993). Ascorbic acid effects on the post-disgorgement oxidative stability of sparkling wine. *American Journal of Enology and Viticulture*, 44, 227–231.
- Mazauric, J. P., & Salmon, J. M. (2005). Interactions between yeast less and wine polyphenols during simulation of wine aging: I. Analyses of remnant polyphenolic compounds in the resulting wines. *Journal of Agricultural and Food Chemistry*, 53, 5647–5653.
- Mazauric, J. P., & Salmon, J. M. (2006). Interactions between yeast lees and wine polyphenols during simulation of wine aging: II. Analysis of desorbed polyphenol compounds from yeast lees. *Journal of Agricultural and Food Chemistry*, 54, 3876–3881.
- Mensor, L. L., Menezes, F. S., Leitão, G. G., Reis, A. S., dos Santos, T. C., Coube, C. S., et al. (2001). Screening of Brazilian plant extracts or antioxidant activity by the use of DPPH<sup>-</sup> free radical method. *Phytotherapy Research*, 15, 127–130.
- OIV Office International de la Vigne et du Vin (2003). HPLC-determination of nine major anthocyanins in red and rosé wine (p. 12). Resolution OENO 22/2003.
- Pezet, R., & Cuenat, Ph. (1996). Resveratrol in wine: Extraction from skin during fermentation and post-fermentation standing of must from gamay grapes. *American Journal of Enology and Viticulture*, 47, 287–290.
- Pozo-Bayón, M. A., Hernández, M. T., Martín Álvares, P. J., & Polo, M. C. (2003). Study of low molecular weight phenolic compounds during the aging of sparkling wines manufactured with red and white grape varieties. *Journal of Agricultural* and Food Chemistry, 51, 2089–2095.
- Ribéreau-Gayón, P., Glories, Y., Maujean, A., & Dubourdieu, D. (2003). Tratado de enología: Microbiologia del vino – vinificaciones y química del vino – estabilización y tratamientos. Buenos Aires: Editorial Hemisferio Sur. p. 1.184.
- Roig, R., Cascón, E., Arola, L., Bladé, C., & Salvadó, M. J. (2002). Procyanidins protect fao cells against hydrogen peroxide-induced oxidative stress. *Biochimica et Biophysica Acta*, 1572, 25–30.
- Satué-Garcia, M. T., Andrés-Lacueva, C., Lamuela-Raventós, R. M., & Frankel, E. N. (1999). Spanish sparkling wines (*Cavas*) as inhibitors of in vitro human lowdensity lipoprotein oxidation. *Journal of Agricultural and Food Chemistry*, 47, 2198–2202.
- Sauer, M., Branduardi, P., Valli, M., & Porro, D. (2004). Production of L-ascorbic acid by metabolically engineered Saccharomyces cerevisiae and Zygosaccharomyces bailii. Applied and Environmental Microbiology, 70, 6086–6091.
- Smirnoff, N., Conklin, P. L., & Loewus, F. A. (2001). Biosyntheses of ascorbic acid in plants: A renaissance. Annual Review of Plant Physiology and Plant Molecular Biology, 52, 437–467.
- Soleas, G. J., Diamandis, E. P., & Goldberg, D. M. (1997). Resveratrol: A molecule whose time has come? And gone? *Clinical Biochemistry*, 30, 91–113.
- Stevens, J. F., Miranda, C. L., Wolthers, K. R., Schimerlik, M., Deinzer, M. L., & Buhler, D. R. (2002). Identification and in vitro biological activities of hop proanthocyanidins: Inhibition of nNOS activity and scavenging of reactive, nitrogen species. *Journal of Agricultural and Food Chemistry*, 50, 3435–3443.
- Threlfall, R., Morris, J. R., & Mauromoustakos, A. (1999). Effects of variety, ultraviolet light exposure, and enological methods on the *trans*-resveratrol level of wine. *American Journal of Enology and Viticulture*, 50, 57–64.
- Valpuesta, V., & Botella, M. A. (2004). Biosynthesis of L-ascorbic acid in plants: New pathways for and old antioxidant. *Trends in Plant Science*, 9, 5773–5777.
- Vrhovsek, U., Wendelin, S., & Eder, R. (1997). Effects of various vinification techniques on the concentration of *cis*-and *trans*-resveratrol and resveratrol glucoside isomers in wine. *American Journal of Enology and Viticulture*, 48, 214–219.
- Yamaguchi, T., Takamura, H., Matoba, T. C., & Terão, J. (1998). HPLC method for evaluation of the free radical-scavenging activity of foods by using 1-1diphenyl-2-picrylhydrazyl. *Bioscience, Biotechnology and Biochemistry*, 62, 1201–1204.
- Yilmaz, Y., & Toledo, R. T. (2004). Major flavonoids in grape seeds and skins: Antioxidant capacity of cathechin, epicathechin, and gallic acid. Journal of Agricultural and Food Chemistry, 52, 255–260.
- Zoecklein, B. W., Fugelsang, K. C., Gump, B. H., & Nury, F. S. (2000). Wine analyses and production. Maryland: Aspen Publishers. p. 613.
- Wilmsen, P. K., Spada, D. S., & Salvador, M. (2005). Antioxidant activity of the flavonoid hesperidin in chemical and biological systems. *Journal of Agricultural* and Food Chemistry, 15, 4757–4761.